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## Chemistry the Great

What would life be,  
Without Chemistry?  
It is simply unthinkable,  
The world would be terrible.

We would not know that atoms make,  
Everything wonderful: a frosted cake.

What is an atom you say?  
I will show you the way...

Well, it originally started with,  
In the century of the fifth,  
Democritus said that atoms weren't hallow,  
But we now know that theory is muy malo.

Then there was Dalton with the next idea,  
Which we know is not true, Dalton, wouldn't wanna be ya.  
Reasons being: there is no internal structure as long  
As there are hooks to bond them together... Wrong!

Here we have Thompson, one creative man,  
Imagined atoms as pastries in a tin can.  
The formula for this was negative particles in positive dough,  
Is this accepted today? Oh no no no!

Almost there, who is next?  
The Planetary Model with orbiting effects.  
Made in 1913,  
But still not correct.

Last but not least, the one we accept today,  
The Quantum Mechanics Model is here to stay.  
Similar to that funky Planetary Model,  
But it still does not orbit or doddle.

Outside the atom lies a special treat,  
Orbiting about 7 days a week.  
The electron is his name,  
And a negative charge is his game.

Past the electron deep inside the atom,  
There is something that we can barely fathom.  
The proton with its positive charge,  
Weighs much more than the electron at large.

Then along came a the neutron with its lovely company,  
Had no charge for anyone to see.  
Proton plus neutrons plus electrons equal atomic mass,  
And now we have our atom at last.

