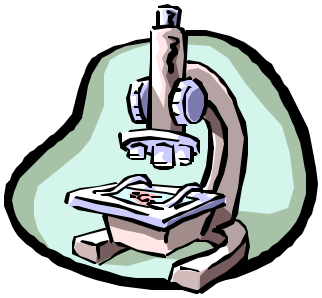
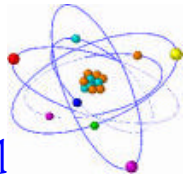
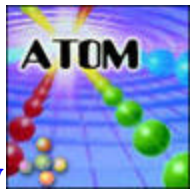


Atoms, atoms all around
Everywhere we look although they don't make a sound
Dalton made an atomic theory
This went down in history
He figured out that all matter is made from particles
Which he discovered from errors and trials
He thought atoms couldn't be made, destroyed or

divided 

However now we know this is misguided
Atoms of different elements have different properties
and a different mass
And for atoms of the same element if you said
Had the same properties and the same mass
I would tell you this was disprov'ed
Electrons all have a negative charge, indeed.
While protons have a positive charge, you see.
And all the while, little neutrons they spin and spin and
spin,
And have a neutral charge, all through out and in.
Now on to more of chemistry
I'll tell you less of Dalton's atomic theory



And more of elements such as mercury
There are more than 100 elements classified in the periodic table
Which you can memorize if you are able
The Kinetic Theory explains solid, liquid, and gas
And yes this does have a bit to do with mass
In the solid state the volume and shape stay the same
With liquids the shape alters but the volume will remain
In the gaseous state the volume and shape alter
Compounds can change from these three states such as water
Now to discuss, compounds and such,
Oh you'll like this my friend, very very much.
Carbon dioxide is CO2,
And calcium hydroxide is Ca(OH)2, whoo hoo!
Now the formula for acetic acid, is quite long indeed,
It spells out HC2 CO3!
Well Ms. Nuno, it is time for the end,
But fear not, for we will soon heal and mend.
See you in class, is what I must say!

So, goodnight and good day!!!